

A New Synthesis of α -Amino Acid Thioesters by Pummerer Reaction of 3-Substituted-4-sulfinyl- β -sultams

Tetsuo Iwama,[†] Tadashi Kataoka,^{*†} Osamu Muraoka,[‡] and Genzoh Tanabe[‡]

Gifu Pharmaceutical University, 6-1, Mitahora-higashi 5-chome, Gifu 502-8585, Japan, and
Kinki University, Faculty of Pharmaceutical Sciences, 3-4-1, Kowakae, Higashi-osaka,
Osaka 577-0818, Japan

Received June 24, 1998

α -Amino acid thioesters were synthesized by the Pummerer reaction of 3-substituted-4-sulfinyl- β -sultams with TFAA. The 3-substituted-4-sulfinyl- β -sultams were prepared from the corresponding β -sultams by sulfonylation with diphenyl disulfide followed by *m*-CPBA oxidation. Diastereoselective synthesis of β -sultams by 1,3-asymmetric induction in [2 + 2] cycloaddition of a sulfene intermediate and chiral imines in solution-phase was studied, and it was found that *N*-alkylimines gave better diastereoselectivities than *N*-aralkylimines. The use of imines derived from (*R*)- and (*S*)- α -methylbenzylamine followed by separation of the major and minor diastereomers gave enantiopure 3-substituted-*N*-methylbenzyl- β -sultams. These β -sultams were then converted to *N*-methylbenzyl- α -amino acid thioesters via sulfonylation and Pummerer rearrangement with high or complete retention of configuration.

Introduction

β -Sultams, sulfonyl analogues of β -lactams, are unique four-membered heterocycles containing three different hetero single bonds, namely, C–N, C–S, and N–S. Stabilization of the β -sultam ring by π -bond overlap between the nitrogen lone-pair electrons and the sulfonyl group is much less than comparable stabilization of the β -lactam ring.¹ In addition, β -sultams are destabilized by the increased distortion of the β -sultam ring due to the C–S and N–S bonds which are longer than the corresponding C–C and C–N bonds of the β -lactam ring. Therefore, β -sultams have been widely investigated from both chemical and pharmacological points of view.² In previous papers we described the hetero bond cleavage of the β -sultam ring.³ If the Pummerer reaction of 3-substituted-4-sulfinyl- β -sultams proceeds successfully, it would become the new method for the C–S bond cleavage of β -sultams and can be utilized for the synthesis of α -amino acid thioesters. α -Amino acid thioesters are activated esters⁴ utilized for peptide synthesis⁵ as well as natural product synthesis as versatile intermediates.⁶ We intended to apply this method to the synthesis

of optically active α -amino acid thioesters⁷ and examined 1,3-asymmetric induction in [2 + 2] cycloaddition of chiral imines and a sulfene intermediate. Synthesis of β -sultams using the [2 + 2] cycloaddition of imines and sulfenes has been given scant attention² since it was first reported by Tsuge and Iwanami.⁸ Recently, Gordeev and co-workers reported the first examples of diastereoselective solid-phase synthesis of β -sultams using chiral imines and activated sulfenes.⁹ In this paper, we report the synthesis of α -amino acid thioesters by the Pummerer reaction of 3-substituted-4-sulfinyl- β -sultams and a preliminary study on diastereoselective synthesis of β -sultams in solution-phase.

Results and Discussion

3-Substituted-4-sulfinyl- β -sultams **4A,B** were synthesized as shown in Scheme 1 (Table 1). β -Sultams

[†] Gifu Pharmaceutical University.

[‡] Kinki University.

(1) Koller, W.; Linkies, A.; Rehling, H.; Reuschling, D. *Tetrahedron Lett.* **1983**, *24*, 2131–2134.

(2) For reviews, see (a) Iwama, T.; Kataoka, T. *Rev. Heteroatom Chem.* **1996**, *15*, 25–60. (b) Harris, P. A. In *Comprehensive Heterocyclic Chemistry II*; Katritzky, A. R., Rees, C. W., Scriven, E. F. V., Eds.; Elsevier Science Inc.: Oxford, 1996; Vol. 1, p 1009. (c) Chanet-Ray, J.; Vessiere, R. *Org. Prep. Proced. Int.* **1986**, *18*, 157–178. (d) Dittmer, D. C.; Sedergran, T. C. In *Small Ring Heterocycles*; Hassner, A., Ed.; John Wiley and Sons: New York, 1985, Part 3, Chapter 5, pp 431–768. (e) Timberlake, J. W.; Elder, E. S. In *Comprehensive Heterocyclic Chemistry*; Katritzky, A. R., Rees, C. W., Eds.; Pergamon Press: Oxford, New York, 1984; Vol. 7, pp 449–489.

(3) (a) Kataoka, T.; Iwama, T. *Tetrahedron Lett.* **1995**, *36*, 245–248. (b) Kataoka, T.; Iwama, T. *Tetrahedron Lett.* **1995**, *36*, 5559–5562. (c) Kataoka, T.; Iwama, T.; Takagi, A. *Tetrahedron Lett.* **1996**, *37*, 2257–2260. (d) Iwama, T.; Takagi, A.; Kataoka, T. *Chem. Pharm. Bull.* **1998**, *46*, 757–766. (e) Iwama, T.; Kataoka, T.; Muraoka, O.; Tanabe, G. *Tetrahedron* **1998**, *54*, 5507–5522.

(4) (a) Wieland, T.; Schäfer, W.; Bokelman, E. *Justus Liebigs Ann. Chem.* **1951**, *573*, 99–105. (b) Farrington, J. A.; Hextall, P. G.; Kenner, G. W.; Turner, J. M. *J. Chem. Soc.* **1957**, 1407–1413.

(5) For some recent papers, see (a) Sasaki, S.; Shinoya, M.; Koga, K. *J. Am. Chem. Soc.* **1985**, *107*, 3371–3372. (b) Sasaki, S.; Koga, K. *Chem. Pharm. Bull.* **1989**, *37*, 912–919. (c) Schnölzer, M.; Kent, S. B. H. *Science* **1992**, *256*, 221–226. (d) Hojo, H.; Know, Y.; Kakuta, Y.; Tsuda, S.; Tanaka, I.; Hikichi, K.; Aimoto, S. *Bull. Chem. Soc. Jpn.* **1993**, *66*, 2700–2706. (e) Richter, L. S.; Tom, J. Y. K.; Burnier, J. P. *Tetrahedron Lett.* **1994**, *35*, 5547–5550. (f) Dawson, P. E.; Muir, T. W.; Clark-Lewis, I.; Kent, S. B. H. *Science* **1994**, *266*, 776–779. (g) Tam, J. P.; Lu, Y. A.; Liu, C. E.; Shao, J. *Proc. Natl. Acad. Sci. U.S.A.* **1995**, *92*, 12485–12489. (h) Hojo, H.; Yoshimura, S.; Go, M.; Aimoto, S. *Bull. Chem. Soc. Jpn.* **1995**, *68*, 330–336. (i) Aimoto, S.; Hojo, H.; Yoshimura, S.; Shimizu, N.; Asano, T.; Ikebe, M.; Kokui, T.; Crivic, A.; Ikura, M. *Pept. Chem.* **1995**, *Volume Date 1994*, *32nd*, 197–200 [Chemical Abstract on CD-ROM, 125: 11415]. (j) Kawakami, T.; Kogure, S.; Aimoto, S. *Bull. Chem. Soc. Jpn.* **1996**, *69*, 3331–3338. (k) Shao, Y.; Lu, W.; Kent, S. B. H. *Tetrahedron Lett.* **1998**, *39*, 3911–3914.

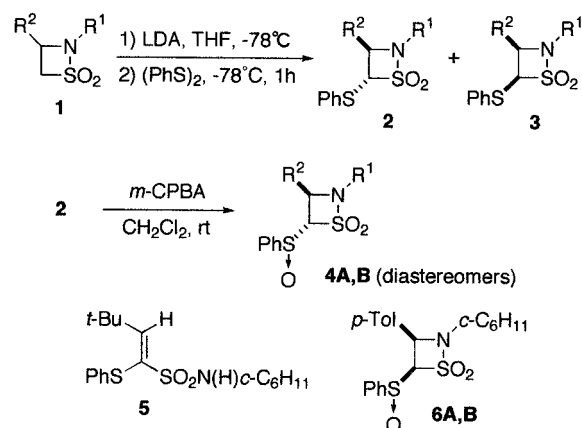
(6) For some recent papers see (a) Fukuyama, T.; Lin, S. C.; Li, L. *J. Am. Chem. Soc.* **1990**, *112*, 7050–7051. (b) Moss, W. O.; Jones, A. C.; Wisedale, R.; Mahon, M. F.; Molloy, K. C.; Bradbury, R. H.; Hales, N. J.; Gallagher, T. *J. Chem. Soc., Perkin Trans. 1* **1992**, 2615–2624. (c) Ho, P. T.; Ngu, K.-y. *J. Org. Chem.* **1993**, *58*, 2313–2316. (d) Jackson, R. F. W.; Palmer, N. J.; Wythes, M. J.; Clegg, W.; Elsegood, M. R. *J. Org. Chem.* **1995**, *60*, 6431–6440.

(7) Williams, R. M. In *Synthesis of Optically Active α -Amino Acids*; Pergamon: New York, 1989.

(8) Tsuge, O.; Iwanami, S. *Bull. Chem. Soc. Jpn.* **1970**, *43*, 3543–3549.

(9) Gordeev, M. F.; Gordon, E. M.; Patel, D. V. *J. Org. Chem.* **1997**, *62*, 8177–8181.

Scheme 1



prepared by [2 + 2] cycloaddition of a sulfene and imines^{3c,d,8} were treated with 3 equiv of LDA at -78 °C in THF followed by 1 equiv of diphenyl disulfide at -78 °C for 1 h to give 3,4-*trans*-4-sulfinyl- β -sultams **2** as major products accompanied by 3,4-*cis*-isomers **3**.¹⁰ The *cis*-isomers **3** were obtained as inseparable mixtures with the starting materials **1** except for **3e**.¹⁰ The use of 1 equiv of LDA generates a mixture of monoanion and dianion of the β -sultams which makes reactions complicated,¹¹ and an excess amount of LDA is required for formation of the sole dianion and for clean reactions. It is necessary to quench reactions at -78 °C; for example, when the reaction of **1f** was quenched at room temperature instead of at -78 °C, a ring-opened product **5** was obtained in 67% yield as a major product together with a trace amount of **2f** (entry 7). Treatment of 3,4-*trans*-4-sulfinyl- β -sultams **2** with *m*-CPBA in CH₂Cl₂ provided 3,4-*trans*-4-sulfinyl- β -sultams **4A,B** as mixtures of stereoisomers at the sulfoxide moiety. Oxidation of 3,4-*cis*-isomer **3d** containing **1d** with *m*-CPBA gave a mixture of four diastereomers, 3,4-*trans*-4-sulfinyl- β -sultams **4dA,dB** and 3,4-*cis*-4-sulfinyl- β -sultams **6A,B**, due to isomerization of the latter to the former.¹²

The Pummerer reaction of 3,4-*trans*-4-sulfinyl- β -sultams **4A,B** with TFAA was examined (Scheme 2, Table 2). Treatment of **4aA** with 1 equiv of TFAA in CH₂Cl₂ at room temperature for 12 h resulted in the recovery of the sulfoxide (entry 1).¹³ A reaction of **4aA** with 2 equiv of TFAA for 24 h gave an α -amino acid thioester **7a** and its amido derivative **8a** in low yields (entry 2).¹³ Thioesters **7** were obtained in good yields accompanied with amides **8** except for **8c,f** regardless of the configuration of the

(10) The stereochemistry was determined from the coupling constants between 3- and 4-protons. In general, *cis*-isomers show greater coupling constants by 1.5–2.0 Hz than *trans*-isomers.^{2a,d} For example, the ¹H NMR spectrum of **2b** showed a couple of doublets at δ 3.86 and 4.69 (owing to 3- and 4-protons) with $J = 6.8$ Hz, and a couple of doublets at δ 4.69 and 5.65, with $J = 8.3$ Hz assignable to 3- and 4-protons of **3b**, were observed in the ¹H NMR spectrum of a mixture of **3b** and **1b**. The *cis*-isomers **3** can be converted to the *trans*-isomers **2** by treatment with LDA in THF at -78 °C.

(11) Müller, M.; Otto, H.-H. *Arch. Pharm. (Weinheim, Ger.)* **1991**, *324*, 15–17.

(12) 3,4-*cis*-4-Sulfinyl- β -sultams **6A,B** gradually isomerized to *trans*-isomers **4dA,dB** during purification with silica gel column chromatography. The ¹H NMR spectrum of the mixture showed four couples of doublets at δ 4.44 and 5.37 with $J = 8.3$ Hz, δ 4.92 and 5.23 with $J = 7.8$ Hz (**6A,B**), δ 4.70 and 4.91 with $J = 4.9$ Hz, and δ 4.24 and 4.85 with $J = 5.4$ Hz (**4dA,dB**) owing to 3- and 4-protons. No *trans*-isomers **4A,B** isomerized to *cis*-isomers with silica gel column chromatography.

(13) The ¹H NMR spectrum of the recovered sulfoxide showed isomerization of the sulfoxide moiety. Two sets of doublet assigned to 3- and 4-protons of **4aA** and **4aB** were observed in the spectrum.

sulfoxide by the use of 4 equiv of TFAA. In the cases of **4c,f**, steric hindrance of the *tert*-butyl group prevented amidation of **7** with excess TFAA. Treatment of a mixture of the four diastereomers **6A,B** and **4dA,B** also provided **7d** and **8d** in 63% and 15% yields, respectively (entry 11). In all cases, a trace amount of diphenyl disulfide was isolated.

A proposed mechanism is shown in Scheme 3. The alcohol **10** is produced by hydrolysis of a Pummerer product **9** formed from **4** and TFAA. Water during workup may cause the hydrolysis. The amide anion **11** is generated by the ring destruction of **10** with the C–S (C–SO₂) bond cleavage followed by elimination of sulfur dioxide to give thioester **7**. A small amount of **8** would be obtained by amidation of **11** with excess TFAA. Diphenyl disulfide is probably obtained by the C–S (C–SPh) bond cleavage with the formation of a β -sultam-4-one **12** although we could not isolate **12**.

Thioesters are activated compounds of carboxylic acids, and α -amino acid thioesters have been utilized for peptide synthesis⁵ as well as natural product synthesis as versatile intermediates.⁶ Therefore, enantioselective synthesis of α -amino acid thioesters is necessary. 1,3-Asymmetric induction was examined in [2 + 2] cycloaddition of a sulfene intermediate and chiral imines **13**, prepared from the corresponding chiral amines and aldehydes (Scheme 4, Table 3). The use of *N*-alkylimines showed moderate diastereoselectivity (entries 1–5), and that of *N*-alkylimines showed good diastereoselectivity (entries 6–8). The best selectivity was given in the case of *N*-(1-*tert*-butylethyl)imine (entry 8). The diastereoselectivity would be dependent on the size and conformation of the *N*-substituents in the imines. Reactions of imines bearing an ether group resulted in the formation of a complex mixture (entries 9 and 10). Diastereomers of compounds **14a–e** bearing an *N*-alkyl group were separable by silica gel column chromatography.

The stereochemistry of the major isomers was determined by X-ray crystallographic analysis of **14h** as a representative (see Supporting Information). The relative configuration is *syn*-form as drawn in Scheme 5. The stereoselectivity in the 1,3-asymmetric induction is explained as follows: a sulfene intermediate, generated from mesyl chloride and an imine, reacts with another imine **15**, which is the most stable conformer by A^{1,3} strain,¹⁴ from the opposite face to the bulky *tert*-butyl group to form a *syn*-isomer.¹³ Contribution of a less stable conformer **16** to the formation of an *anti*-isomer would be negligible. The stereochemistry of the major of **14a,b** is deduced to be *syn* from this mechanism.

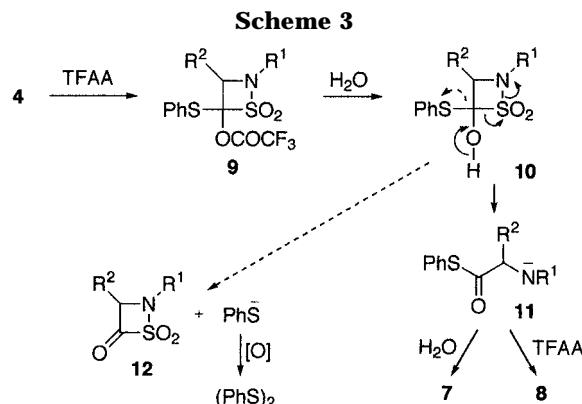
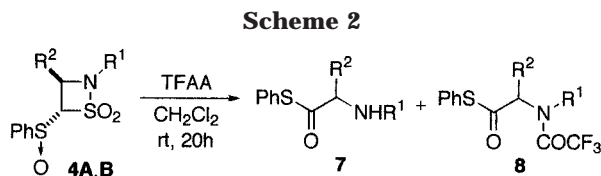
We applied this thioester synthesis to chiral β -sultams prepared by 1,3-asymmetric induction described above (Scheme 6). Sulfoxides **19** and **23** were prepared by sulfinylation of **14** followed by *m*-CPBA oxidation of *trans*-isomers **17** and **21**, respectively. The sulfoxides **19A,B** and **23A,B** were obtained as two separable stereoisomers at sulfoxide group (alphabets **A** and **B** mean stereoisomers at sulfoxide moiety). Treatment of **19** and **23** with 4 equiv of TFAA gave chiral α -amino acid thioesters **20** and **24**, respectively, in high yields. Slight epimerization of the α -chiral center of α -phenyl thioesters **20a** and **24a** was observed under the reaction conditions, and the diastereomeric excess was >90%, determined by the ¹H NMR spectra.¹⁵ No epimerization was observed

(14) Hoffmann, R. W. *Chem. Rev.* **1989**, *89*, 1841–1860.

Table 1. Synthesis of 3-Substituted-4-sulfinyl- β -sultams

entry	1	R ¹	R ²	2 and 3 (%yield) ^a	4 ^b (%yield) ^a
1	1a	<i>c</i> -C ₆ H ₁₁	Ph	2a (72) 3a (trace) ^c	4aA (58), 4aB (16)
2	1b	<i>n</i> -Bu	Ph	2b (57) 3b (12) ^c	4bA (49), 4bB (47)
3	1c	<i>t</i> -Bu	Ph	2c (67) 3c (7) ^c	4cA (75), 4cB (20)
4	1d	<i>c</i> -C ₆ H ₁₁	<i>p</i> -tolyl	2d (66) 3d (18) ^c	4dA (53), 4dB (20)
5	1e	<i>c</i> -C ₆ H ₁₁	<i>o</i> -tolyl	2e (72) 3e (18)	4eA (64), 4eB (33)
6	1f	<i>c</i> -C ₆ H ₁₁	<i>t</i> -Bu	2f (92)	4fA (54), 4fB (38)
7 ^d	1f	<i>c</i> -C ₆ H ₁₁	<i>t</i> -Bu	2f (t), 5 (67)	

^a Isolated yield unless otherwise mentioned. ^b A,B: Diastereomers at sulfoxide moiety. ^c The compound was obtained as an inseparable mixture with the starting material. The yield was estimated by the ¹H NMR spectrum. ^d The reaction was quenched at rt. ^e t: Trace.

**Table 2. The Pummerer Reaction of 3-Substituted-4-sulfinyl- β -sultams**

entry	4	R ¹	R ²	TFAA (equiv)	Products ^a (%yield) ^b
1 ^c	4aA	<i>c</i> -C ₆ H ₁₁	Ph	1.0	recovery ^d
2 ^e	4aA	<i>c</i> -C ₆ H ₁₁	Ph	2.0	7a , 8a (low yield) ^f
3	4aA	<i>c</i> -C ₆ H ₁₁	Ph	4.0	7a (72), 8a (13)
4	4aB	<i>c</i> -C ₆ H ₁₁	Ph	4.0	7a (79), 8a (14)
5	4bA	<i>n</i> -Bu	Ph	4.0	7b (74), 8b (10)
6	4bB	<i>n</i> -Bu	Ph	4.0	7b (73), 8b (9)
7	4cA	<i>t</i> -Bu	Ph	4.0	7c (93)
8	4cB	<i>t</i> -Bu	Ph	4.0	7c (97)
9	4dA	<i>c</i> -C ₆ H ₁₁	<i>p</i> -tolyl	4.0	7d (73), 8d (9)
10	4dB	<i>c</i> -C ₆ H ₁₁	<i>p</i> -tolyl	4.0	7d (79), 8d (8)
11	4d,6^g	<i>c</i> -C ₆ H ₁₁	<i>p</i> -tolyl	4.0	7d (63), 8d (15)
12	4eA	<i>c</i> -C ₆ H ₁₁	<i>o</i> -tolyl	4.0	7e (71), 8e (10)
13	4eB	<i>c</i> -C ₆ H ₁₁	<i>o</i> -tolyl	4.0	7e (73), 8e (7)
14	4fA	<i>c</i> -C ₆ H ₁₁	<i>t</i> -Bu	4.0	7f (81)
15	4fB	<i>c</i> -C ₆ H ₁₁	<i>t</i> -Bu	4.0	7f (82)

^a A trace amount of (PhS)₂ was isolated in all cases. ^b Isolated yield. ^c Reaction time: 12 h. ^d Isomerization of the sulfoxide moiety was observed. ^e Reaction time: 24 h. ^f Yields of **7a** and **8a** were not calculated. A considerable amount of the starting material was recovered. Isomerization of the sulfoxide moiety was observed. ^g Mixture of 3,4-*cis*-**6A,B** and 3,4-*trans*- β -sultams **4dA,B** was used.

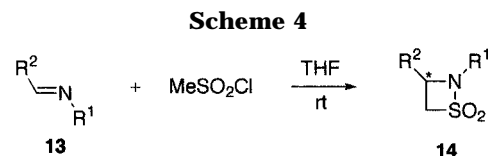
in the ¹H NMR spectra in the cases of α -*tert*-butyl thioesters **20b** and **24b**.

Experimental Section

General Methods. Melting points are uncorrected. IR spectra were recorded by KBr (solids) and NaCl (liquids) methods. ¹H NMR spectra were recorded in a CDCl₃ solution with tetramethylsilane as an internal standard. ¹³C NMR spectra were obtained with CDCl₃ (77.0 ppm) as an internal standard. Mass spectra were recorded with a direct-insertion probe at 70 eV. All chromatographic isolations were accomplished with either Kieselgel 60 (Merck) or BW-127ZH (Fuji Silysia) for column chromatography or Kieselgel 60 PF₂₅₄ containing gypsum (Merck) for preparative TLC.

Synthesis of β -Sultams **1 and **14** by [2 + 2] Cycloaddition.** β -Sultams **1** and **14** were synthesized according to the literature.^{3d,e}

(15) In prolonged reaction or prolonged workup, the diastereomeric excess decreased to ca. 75% in the worst case. Rapid workup and purification are necessary to minimize epimerization.



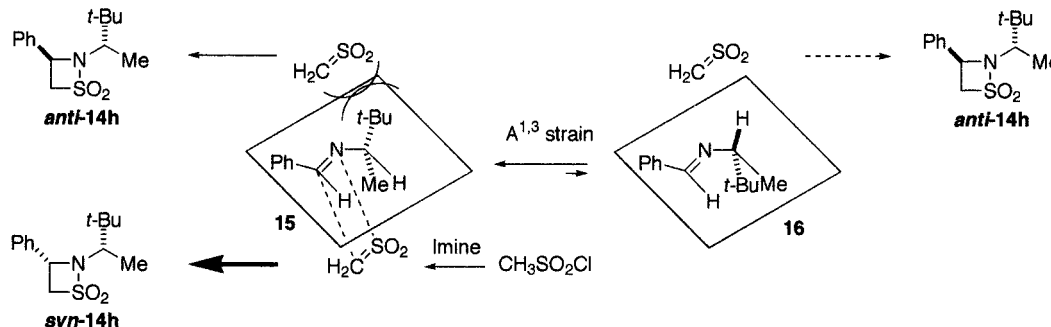
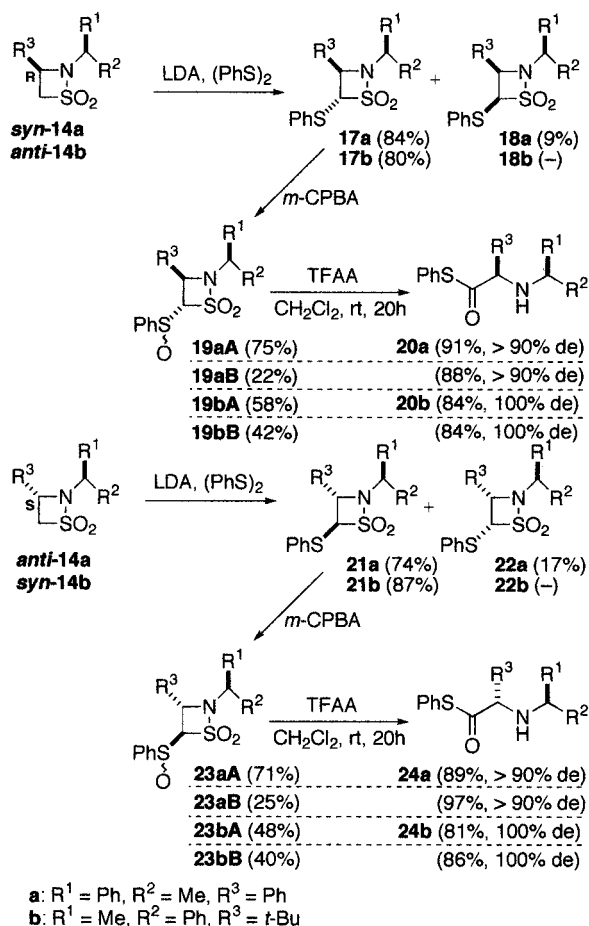
(3*R*,1'*R*)-2-(α -Methylbenzyl)-3-phenyl-1,2-thiazetidine 1,1-dioxide (*syn*-14a**):** colorless prisms (from CH₂Cl₂-hexane), mp 131–133 °C; ¹H NMR (CDCl₃) δ : 1.60 (3 H, d, *J* = 7 Hz), 3.93–4.00 (2 H, m), 4.29–4.36 (1 H, m), 4.48 (1 H, q, *J* = 7 Hz), 6.97–7.15 (10 H, m); ¹³C NMR (CDCl₃) δ : 21.7, 50.3, 58.9, 65.6, 126.8, 128.0, 128.2, 128.4, 137.6, 138.1, an aromatic carbon is overlapped; MS (FAB) *m/z* (rel int %): 288 (100, M⁺ + 1); IR ν_{\max} (KBr) cm⁻¹: 1325, 1140 (SO₂). Anal. Calcd for C₁₆H₁₇NO₂S: C, 66.87; H, 5.96; N, 4.87. Found: C, 66.65; H, 5.98; N, 4.87.

(3*S*,1'*R*)-2-(α -Methylbenzyl)-3-phenyl-1,2-thiazetidine 1,1-dioxide (*anti*-14a**):** colorless prisms (from EtOAc-hexane), mp 105–108 °C; ¹H NMR (CDCl₃) δ : 1.49 (3 H, d, *J* = 6.8 Hz), 3.94 (1 H, dd, *J* = 5 and 12.2 Hz), 4.26 (1 H, dd, *J* = 5 and 7.8 Hz), 4.30 (1 H, q, *J* = 6.8 Hz), 4.39 (1 H, dd, *J* = 7.8 and 12.2 Hz), 7.24–7.44 (10 H, m); ¹³C NMR (CDCl₃) δ : 20.9, 49.1, 57.4, 65.8, 126.9, 127.3, 128.0, 128.6, 128.9, 129.0, 137.8, 140.2; MS (FAB) *m/z* (rel int %): 288 (67, M⁺ + 1), 105 (100); IR ν_{\max} (KBr) cm⁻¹: 1315, 1145 (SO₂). Anal. Calcd for C₁₆H₁₇NO₂S: C, 66.87; H, 5.96; N, 4.87. Found: C, 66.71; H, 5.94; N, 4.94.

Table 3. 1,3-Asymmetric Induction in the [2 + 2] Cycloaddition of a Sulfene Intermediate and Chiral Imines^a

entry	R ¹	R ²	14 (%Yield) ^b
1	(<i>R</i>)- α -methylbenzyl	Ph	14a , (70, 42% de) ^c
2	(<i>S</i>)- α -methylbenzyl	<i>t</i> -Bu	14b , (32, 45% de) ^c
3	(<i>R</i>)- α ,4-dimethylbenzyl	Ph	14c , (72, 44% de) ^c
4	rac-1-(1-naphthyl)ethyl	Ph	14d , (53, 47% de) ^c
5	rac-1-indanyl	Ph	14e , (36, 50% de) ^c
6	rac-1-cyclohexylethyl	Ph	14f , (60, 67% de) ^d
7	(1 <i>R</i> ,2 <i>R</i> ,3 <i>R</i> ,5 <i>S</i>)-isopinocampheyl	Ph	14g , (54, 80% de) ^d
8	rac-1- <i>tert</i> -butylethyl	Ph	14h , (67, >95% de) ^d
9	rac-1-(methoxymethyl)propyl	Ph	complex mixture
10	(<i>R</i>)-2-(methoxymethyl)pyrrolidinyl	Ph	complex mixture

^a 2 equiv of imines were used based on MsCl. ^b Isolated yield based on MsCl. Diastereomeric excess was calculated by the ¹H NMR spectrum of the reaction mixture. ^c Separable stereoisomers. ^d Inseparable stereoisomers.

Scheme 5**Scheme 6**

(3*S*,1'*S*)-3-*tert*-Butyl-2-(α -methylbenzyl)-1,2-thiazetidine 1,1-dioxide (*syn*-14b): colorless leaves (from EtOAc–hexane), mp 148–152 °C; ¹H NMR (CDCl₃) δ : 0.85 (9 H, s), 1.76 (3 H, d, J = 7 Hz), 3.27 (1 H, dd, J = 5.9 and 8.3 Hz), 3.76 (1 H, dd, J = 5.9 and 12.2 Hz), 3.94 (1 H, dd, J = 8.3 and

12.2 Hz), 4.50 (1 H, q, J = 7 Hz), 7.30 (1 H, t, J = 7.3 Hz), 7.37 (2 H, t, J = 7.3 Hz), 7.47 (2 H, d, J = 7.3 Hz); ¹³C NMR (CDCl₃) δ : 19.0, 25.8, 33.6, 55.5, 58.4, 58.7, 127.8, 127.9, 128.5, 140.2; MS (EI) m/z (rel int %): 267 (1, M⁺), 105 (100); IR ν_{\max} (KBr) cm⁻¹: 1315, 1150 (SO₂). Anal. Calcd for C₁₄H₂₁NO₂S: C, 62.89; H, 7.92; N, 5.24. Found: C, 62.62; H, 8.06; N, 5.18.

(3*R*,1'*S*)-3-*tert*-Butyl-2-(α -methylbenzyl)-1,2-thiazetidine 1,1-dioxide (*anti*-14b): colorless leaves (from EtOAc–hexane), mp 105–109 °C; ¹H NMR (CDCl₃) δ : 0.87 (9 H, s), 1.81 (3 H, d, J = 7 Hz), 3.10 (1 H, dd, J = 5 and 8 Hz), 3.78 (1 H, dd, J = 5 and 12 Hz), 3.94 (1 H, dd, J = 8 and 12 Hz), 4.51 (1 H, q, J = 7 Hz), 7.30 (1 H, t, J = 7 Hz), 7.38 (2 H, t, J = 7 Hz), 7.49 (2 H, d, J = 7 Hz); ¹³C NMR (CDCl₃) δ : 20.0, 25.9, 33.8, 53.0, 58.3, 58.8, 127.7, 127.7, 128.7, 140.4; MS (EI) m/z (rel int %): 267 (3, M⁺), 105 (100); IR ν_{\max} (KBr) cm⁻¹: 1295, 1145 (SO₂). Anal. Calcd for C₁₄H₂₁NO₂S: C, 62.89; H, 7.92; N, 5.24. Found: C, 62.61; H, 8.04; N, 5.17.

(3*R,1'*R**)-2-(1-*tert*-Butylethyl)-3-phenyl-1,2-thiazetidine 1,1-dioxide (*syn*-14h):** colorless needles (from EtOAc–hexane), mp 148–151 °C; ¹H NMR (CDCl₃) δ : 0.99 (9 H, s), 1.35 (3 H, d, J = 6.8 Hz), 2.58 (1 H, q, J = 6.8 Hz), 3.91 (1 H, dd, J = 5.9 and 12 Hz), 4.31 (1 H, dd, J = 7.8 and 12 Hz), 4.39 (1 H, dd, J = 5.9 and 7.8 Hz), 7.37–7.43 (3 H, m), 7.49 (2 H, d, J = 7.8 Hz); ¹³C NMR (CDCl₃) δ : 11.6, 27.6, 34.5, 48.4, 64.8, 65.3, 127.7, 129.3, 129.5, 137.7; MS (EI) m/z (rel int %): 267 (1, M⁺), 210 (100); IR ν_{\max} (KBr) cm⁻¹: 1305, 1140 (SO₂). Anal. Calcd for C₁₄H₂₁NO₂S: C, 62.89; H, 7.92; N, 5.24. Found: C, 62.62; H, 7.90; N, 5.19.

Sulfenylation of β -Sultams 1 and 14. General Procedure. To a solution of LDA (3 mmol, prepared from 3 mmol of diisopropylamine (0.39 cm³) and 3 mmol of ⁿBuLi in hexane) in dry THF (10 cm³) was added dropwise a solution of β -sultam 1 or 14 (1 mmol) in THF (2–4 cm³) at –78 °C under argon. After 30 min, diphenyl disulfide (218 mg, 1 mmol) in THF (1–2 cm³) was added dropwise to it, and the whole was stirred at –78 °C for 1 h. The reaction was quenched by addition of saturated aqueous NH₄Cl (0.5 cm³) at –78 °C, and the mixture was warmed to room temperature. Additional saturated aqueous NH₄Cl (5 cm³) and water (5 cm³) were added to it, and the organic layer was separated. The water layer was extracted twice with EtOAc (10 cm³). The organic layer and the extracts were combined, washed with saturated aqueous NaCl (20 cm³), dried (MgSO₄), and evaporated under reduced

pressure. The residue was purified by column chromatography on silica gel with EtOAc–hexane (1:20–1:10 v/v).

trans-2-Cyclohexyl-4-(phenylsulfenyl)-3-*p*-tolyl-1,2-thiazetidine 1,1-dioxide (2d): colorless prisms (from EtOAc–hexane), mp 102–104 °C; $^1\text{H NMR}$ (CDCl_3) δ : 1.11–1.26 (4 H, m), 1.47–1.61 (4 H, m), 1.72 (1 H, br d, $J = 13$ Hz), 2.03 (1 H, br d, $J = 12$ Hz), 2.37 (3 H, s), 3.21 (1 H, m), 3.96 (1 H, d, $J = 6.3$ Hz), 4.97 (1 H, d, $J = 6.3$ Hz), 7.21 (2 H, d, $J = 8$ Hz), 7.26–7.31 (3 H, m), 7.38–7.41 (4 H, m); $^{13}\text{C NMR}$ (CDCl_3) δ : 21.2, 24.2, 24.4, 25.3, 30.4, 31.9, 56.8, 57.5, 82.5, 126.5, 128.4, 129.4, 129.7, 131.6, 131.8, 134.1, 139.1; MS (FAB) m/z (rel int %): 388 (7, $\text{M}^+ + 1$), 154 (100); IR ν_{max} (KBr) cm^{-1} : 1310, 1155 (SO_2). Anal. Calcd for $\text{C}_{21}\text{H}_{25}\text{NO}_2\text{S}_2$: C, 65.08; H, 6.50; N, 3.61. Found: C, 64.91; H, 6.57; N, 3.60.

trans-3-*tert*-Butyl-2-cyclohexyl-4-(phenylsulfenyl)-1,2-thiazetidine 1,1-dioxide (2f): colorless prisms (from CH_2Cl_2 –hexane), mp 106–108 °C; $^1\text{H NMR}$ (CDCl_3) δ : 1.05 (9 H, s), 1.02–1.25 (3 H, m), 1.58–1.87 (5 H, m), 1.98 (1 H, br d, $J = 13$ Hz), 2.22 (1 H, br d, $J = 12$ Hz), 3.01–3.09 (1 H, m), 3.10 (1 H, d, $J = 5.9$ Hz), 4.89 (1 H, d, $J = 5.9$ Hz), 7.30–7.38 (3 H, m), 7.50 (2 H, dd, $J = 2$ and 8 Hz); $^{13}\text{C NMR}$ (CDCl_3) δ : 25.2, 25.4, 26.2, 26.5, 28.6, 31.4, 34.6, 58.7, 59.1, 75.5, 128.5, 129.5, 132.0, 132.1; MS (FAB) m/z (rel int %): 354 (2, $\text{M}^+ + 1$), 290 (100); IR ν_{max} (KBr) cm^{-1} : 1305, 1165 (SO_2). Anal. Calcd for $\text{C}_{18}\text{H}_{27}\text{NO}_2\text{S}_2$: C, 61.15; H, 7.70; N, 3.96. Found: C, 60.97; H, 7.68; N, 3.87.

2-*tert*-Butyl-*N*-cyclohexyl-1-phenylsulfenylvinylsulfonamide (5): colorless prisms (from EtOAc–hexane), mp 104–110 °C; $^1\text{H NMR}$ (CDCl_3) δ : 1.11–1.33 (5 H, m), 1.27 (9 H, s), 1.53–1.58 (1 H, m), 1.66–1.71 (2 H, m), 1.81–1.84 (2 H, m), 3.03–3.07 (1 H, m), 4.43 (1 H, br d, $J = 7.3$ Hz), 7.15–7.33 (5 H, m), 7.62 (1 H, s); $^{13}\text{C NMR}$ (CDCl_3) δ : 24.9, 25.4, 29.9, 34.2, 35.6, 53.4, 126.5, 127.4, 129.3, 131.8, 135.4, 162.2; MS (EI) m/z (rel int %): 353 (27, M^+), 191 (100); IR ν_{max} (KBr) cm^{-1} : 3270 (NH), 1320, 1155 (SO_2). Anal. Calcd for $\text{C}_{18}\text{H}_{27}\text{NO}_2\text{S}_2$: C, 61.15; H, 7.70; N, 3.96. Found: C, 61.09; H, 7.74; N, 3.99.

(3*R*,4*S*,1'*R*')-2-(α -Methylbenzyl)-3-phenyl-4-(phenylsulfenyl)-1,2-thiazetidine 1,1-dioxide (17a): colorless prisms (from CH_2Cl_2 –hexane), mp 96–98 °C; $^1\text{H NMR}$ (CDCl_3) δ : 1.65 (3 H, d, $J = 7$ Hz), 3.62 (1 H, d, $J = 6.8$ Hz), 4.45 (1 H, q, $J = 7$ Hz), 5.07 (1 H, d, $J = 6.8$ Hz), 6.96–7.16 (10 H, m), 7.28–7.30 (3 H, m), 7.37–7.39 (2 H, m); $^{13}\text{C NMR}$ (CDCl_3) δ : 21.8, 58.2, 59.3, 82.4, 127.0, 128.1, 128.1, 128.3, 128.4, 128.5, 128.6, 129.5, 131.3, 132.1, 136.2, 137.2; MS (FAB) m/z (rel int %): 396 (2, $\text{M}^+ + 1$), 105 (100); IR ν_{max} (KBr) cm^{-1} : 1320, 1160 (SO_2). Anal. Calcd for $\text{C}_{22}\text{H}_{21}\text{NO}_2\text{S}_2$: C, 66.81; H, 5.35; N, 3.54. Found: C, 66.73; H, 5.39; N, 3.53.

(3*R*,4*R*,1'*R*')-2-(α -Methylbenzyl)-3-phenyl-4-(phenylsulfenyl)-1,2-thiazetidine 1,1-dioxide (18a): colorless needles (from CH_2Cl_2 –hexane), mp 151–154 °C; $^1\text{H NMR}$ (CDCl_3) δ : 1.63 (3 H, d, $J = 7$ Hz), 4.45 (1 H, d, $J = 9.2$ Hz), 4.63 (1 H, q, $J = 7$ Hz), 5.60 (1 H, d, $J = 9.2$ Hz), 7.04–7.33 (15 H, m); $^{13}\text{C NMR}$ (CDCl_3) δ : 21.3, 54.9, 57.9, 81.1, 128.0, 128.2, 128.4, 128.8, 129.4, 131.8, 132.1, 134.3, 137.7, three aromatic carbons are overlapped; MS (FAB) m/z (rel int %): 396 (3, $\text{M}^+ + 1$), 105 (100); IR ν_{max} (KBr) cm^{-1} : 1310, 1165 (SO_2). Anal. Calcd for $\text{C}_{22}\text{H}_{21}\text{NO}_2\text{S}_2$: C, 66.81; H, 5.35; N, 3.54. Found: C, 66.68; H, 5.50; N, 3.37.

(3*S*,4*R*,1'*R*')-2-(α -Methylbenzyl)-3-phenyl-4-(phenylsulfenyl)-1,2-thiazetidine 1,1-dioxide (21a): colorless oil; $^1\text{H NMR}$ (CDCl_3) δ : 1.49 (3 H, d, $J = 7$ Hz), 3.90 (1 H, d, $J = 5.9$ Hz), 4.31 (1 H, q, $J = 7$ Hz), 5.08 (1 H, d, $J = 5.9$ Hz), 7.24–7.44 (15 H, m); $^{13}\text{C NMR}$ (CDCl_3) δ : 21.0, 57.1, 57.9, 82.7, 126.9, 127.3, 128.1, 128.7, 128.7, 129.2, 129.3, 129.5, 131.1, 132.4, 136.1, 139.8; MS (FAB) m/z (rel int %): 396 (2, $\text{M}^+ + 1$), 154 (100); IR ν_{max} (KBr) cm^{-1} : 1320, 1165 (SO_2). Anal. Calcd for $\text{C}_{22}\text{H}_{21}\text{NO}_2\text{S}_2$: C, 66.81; H, 5.35; N, 3.54. Found: C, 66.64; H, 5.46; N, 3.36.

(3*S*,4*S*,1'*R*')-2-(α -Methylbenzyl)-3-phenyl-4-(phenylsulfenyl)-1,2-thiazetidine 1,1-dioxide (22a): colorless prisms (from CH_2Cl_2 –hexane), mp 95–97 °C; $^1\text{H NMR}$ (CDCl_3) δ : 1.63 (3 H, d, $J = 7$ Hz), 4.38 (1 H, q, $J = 7$ Hz), 4.62 (1 H, d, $J = 8.3$ Hz), 5.64 (1 H, d, $J = 8.3$ Hz), 7.26–7.36 (10 H, m), 7.42 (5 H, s); $^{13}\text{C NMR}$ (CDCl_3) δ : 21.2, 54.6, 57.0, 81.0, 127.2, 128.1,

128.3, 128.5, 128.6, 128.7, 129.5, 131.8, 132.2, 133.5, 140.1, an aromatic carbon is overlapped; MS (FAB) m/z (rel int %): 396 (2, $\text{M}^+ + 1$), 154 (100); IR ν_{max} (KBr) cm^{-1} : 1315, 1160 (SO_2). Anal. Calcd for $\text{C}_{22}\text{H}_{21}\text{NO}_2\text{S}_2$: C, 66.81; H, 5.35; N, 3.54. Found: C, 66.64; H, 5.46; N, 3.40.

Synthesis of 4-Sulfinyl- β -sultams 4, 19, and 23. General Procedure. To a solution of a 4-sulfinyl- β -sultam 2, 17, or 21 (0.5 mmol) in CH_2Cl_2 (10–20 cm^3) was added *m*CPBA (70–85% purity, 0.5 mmol) at 0 °C. After 1–3 h, saturated aqueous NaHCO_3 (10 cm^3) was added to it, and the organic layer was separated. The water layer was extracted once with CH_2Cl_2 (10 cm^3). The organic layer and the extracts were combined, washed with saturated aqueous NaCl (10 cm^3), dried (MgSO_4), and evaporated under reduced pressure. The residue was purified by column chromatography on silica gel with EtOAc–hexane (1:10–1:3 v/v).

trans-2-Cyclohexyl-4-(phenylsulfenyl)-3-*p*-tolyl-1,2-thiazetidine 1,1-Dioxide (4d). Major isomer **4dA**: colorless prisms (from EtOAc–hexane), mp 161–164 °C; $^1\text{H NMR}$ (CDCl_3) δ : 1.11–1.29 (4 H, m), 1.43–1.77 (5 H, m), 2.01 (1 H, br d, $J = 12$ Hz), 2.29 (3 H, s, Me), 3.31–3.37 (1 H, m), 4.70 (1 H, d, $J = 4.9$ Hz), 4.91 (1 H, d, $J = 4.9$ Hz), 7.05 (2 H, d, $J = 7.8$ Hz), 7.22 (2 H, d, $J = 7.8$ Hz), 7.48–4.79 (3 H, m), 7.67–6.69 (2 H, m); $^{13}\text{C NMR}$ (CDCl_3) δ : 21.1, 24.1, 24.3, 25.3, 30.4, 31.8, 48.5, 56.8, 92.5, 124.2, 126.6, 129.6, 129.7, 132.2, 134.3, 138.8, 140.4; MS (FAB) m/z (rel int %): 404 (12, $\text{M}^+ + 1$), 154 (100); IR ν_{max} (KBr) cm^{-1} : 1320, 1165 (SO_2), 1055 (SO). Anal. Calcd for $\text{C}_{21}\text{H}_{25}\text{NO}_3\text{S}_2$: C, 62.50; H, 6.24; N, 3.47. Found: C, 62.30; H, 6.30; N, 3.30.

Minor isomer **4dB**: colorless prisms (from EtOAc–hexane), mp 146–151 °C; $^1\text{H NMR}$ (CDCl_3) δ : 1.06–1.27 (4 H, m), 1.39–1.69 (5 H, m), 1.96 (1 H, br d, $J = 12$ Hz), 2.30 (3 H, s), 3.25–3.30 (1 H, m), 4.24 (1 H, d, $J = 5.4$ Hz), 4.85 (1 H, d, $J = 5.4$ Hz), 7.01–7.03 (4 H, m), 7.48–4.57 (3 H, m), 7.73 (2 H, dd, $J = 1.5$ and 8 Hz); $^{13}\text{C NMR}$ (CDCl_3) δ : 21.1, 23.9, 24.1, 25.2, 30.2, 31.7, 50.7, 56.9, 93.2, 125.6, 126.3, 129.6, 129.7, 132.8, 133.4, 139.0, 139.0; MS (FAB) m/z (rel int %): 404 (16, $\text{M}^+ + 1$), 154 (100); IR ν_{max} (KBr) cm^{-1} : 1320, 1165 (SO_2), 1060 (SO). Anal. Calcd for $\text{C}_{21}\text{H}_{25}\text{NO}_3\text{S}_2$: C, 62.50; H, 6.24; N, 3.47. Found: C, 62.27; H, 6.27; N, 3.32.

trans-3-*tert*-Butyl-2-cyclohexyl-4-(phenylsulfenyl)-1,2-thiazetidine 1,1-Dioxide (4f). Major isomer **4fA**: colorless needles (from CH_2Cl_2 –hexane), mp 187–191 °C; $^1\text{H NMR}$ (CDCl_3) δ : 0.85 (9 H, s), 1.19–1.26 (3 H, m), 1.62–1.85 (5 H, m), 2.18 (2 H, br d, $J = 12$ Hz), 3.13–3.20 (1 H, m), 3.87 (1 H, d, $J = 3.9$ Hz), 4.59 (1 H, d, $J = 3.9$ Hz), 7.56–7.60 (3 H, m), 7.74–7.76 (2 H, m); $^{13}\text{C NMR}$ (CDCl_3) δ : 25.3, 25.4, 26.1, 26.3, 29.9, 31.4, 33.9, 55.0, 59.2, 88.2, 124.4, 129.7, 132.1, 141.3; MS (FAB) m/z (rel int %): 370 (25, $\text{M}^+ + 1$), 154 (100); IR ν_{max} (KBr) cm^{-1} : 1310, 1160 (SO_2), 1055 (SO). Anal. Calcd for $\text{C}_{18}\text{H}_{27}\text{NO}_3\text{S}_2$: C, 58.50; H, 7.36; N, 3.79. Found: C, 58.21; H, 7.41; N, 3.71.

Minor isomer **4fB**: colorless needles (from CH_2Cl_2 –hexane), mp 185–187 °C; $^1\text{H NMR}$ (CDCl_3) δ : 0.96 (9 H, s), 1.13–1.23 (3 H, m), 1.64–1.82 (5 H, m), 2.00 (1 H, br d, $J = 13$ Hz), 2.13 (1 H, br d, $J = 12$ Hz), 3.05–3.11 (1 H, m), 3.59 (1 H, d, $J = 4.9$ Hz), 4.68 (1 H, d, $J = 4.9$ Hz), 7.58–7.59 (3 H, m), 7.79–7.82 (2 H, m); $^{13}\text{C NMR}$ (CDCl_3) δ : 25.3, 25.4, 26.1, 26.2, 29.4, 31.4, 34.4, 54.6, 59.3, 86.0, 125.5, 129.6, 132.4, 139.2; MS (FAB) m/z (rel int %): 370 (34, $\text{M}^+ + 1$), 154 (100); IR ν_{max} (KBr) cm^{-1} : 1315, 1175 (SO_2), 1055 (SO). Anal. Calcd for $\text{C}_{18}\text{H}_{27}\text{NO}_3\text{S}_2$: C, 58.50; H, 7.36; N, 3.79. Found: C, 58.22; H, 7.36; N, 3.65.

(3*R*,4*S*,1'*R*')-2-(α -Methylbenzyl)-3-phenyl-4-phenylsulfenyl-1,2-thiazetidine 1,1-Dioxide (19a). Major isomer **19aA**: colorless prisms (from CH_2Cl_2 –hexane), mp 139–141 °C; $^1\text{H NMR}$ (CDCl_3) δ : 1.62 (3 H, d, $J = 6.3$ Hz), 4.57 (1 H, d, $J = 5$ Hz), 4.61 (1 H, q, $J = 6.3$ Hz), 4.78 (1 H, d, $J = 5$ Hz), 6.94 (2 H, d, $J = 7$ Hz), 7.01–7.08 (6 H, m), 7.12 (2 H, d, $J = 7$ Hz), 7.39–7.40 (3 H, m), 7.57–7.60 (2 H, m); $^{13}\text{C NMR}$ (CDCl_3) δ : 21.9, 49.2, 58.8, 92.5, 124.1, 126.9, 128.2, 128.3, 128.3, 129.6, 132.1, 137.0, 137.3, 140.0, two aromatic carbons are overlapped; MS (FAB) m/z (rel int %): 412 (13, $\text{M}^+ + 1$), 125 (100); IR ν_{max} (KBr) cm^{-1} : 1330, 1165 (SO_2), 1045 (SO).

Anal. Calcd for $C_{22}H_{21}NO_3S_2$: C, 64.21; H, 5.14; N, 3.40. Found: C, 64.11; H, 5.17; N, 3.40.

Minor isomer **19aB**: colorless prisms (from CH_2Cl_2 -hexane), mp 199–202 °C; 1H NMR ($CDCl_3$) δ : 1.59 (3 H, d, $J = 6.8$ Hz), 3.88 (1 H, d, $J = 6$ Hz), 4.54 (1 H, q, $J = 6.8$ Hz), 4.94 (1 H, d, $J = 6$ Hz), 6.74 (2 H, d, $J = 7$ Hz), 6.95–7.08 (8 H, m), 7.40–7.48 (3 H, m), 7.65 (2 H, dd, $J = 1$ and 8 Hz); ^{13}C NMR ($CDCl_3$) δ : 21.9, 52.2, 59.1, 93.2, 125.4, 126.7, 128.1, 128.3, 128.3, 128.4, 129.7, 132.8, 135.8, 137.0, 138.9, an aromatic carbon is overlapped; MS (FAB) m/z (rel int %): 412 (39, $M^+ + 1$), 125 (100); IR ν_{max} (KBr) cm^{-1} : 1300, 1165 (SO_2), 1060 (SO). Anal. Calcd for $C_{22}H_{21}NO_3S_2$: C, 64.21; H, 5.14; N, 3.40. Found: C, 63.94; H, 5.17; N, 3.38.

(3S,4R,1'R)-2-(α -Methylbenzyl)-3-phenyl-4-(phenylsulfinyl)-1,2-thiazetidene 1,1-Dioxide (23a). Major isomer **23aA**: white powder (from EtOAc-hexane), mp 160–162 °C; 1H NMR ($CDCl_3$) δ : 1.53 (3 H, d, $J = 6.8$ Hz), 4.41 (1 H, q, $J = 6.8$ Hz), 4.80 and 4.82 (each 1 H, d, $J = 4$ Hz), 7.25–7.34 (10 H, m), 7.46–7.48 (3 H, m), 7.66–7.68 (2 H, m); ^{13}C NMR ($CDCl_3$) δ : 21.4, 49.1, 57.5, 92.8, 124.2, 125.7, 127.0, 127.1, 128.2, 128.8, 129.0, 129.7, 132.2, 136.0, 139.8, 140.3; MS (FAB) m/z (rel int %): 412 (23, $M^+ + 1$), 154 (100); IR ν_{max} (KBr) cm^{-1} : 1315, 1165 (SO_2), 1045 (SO). Anal. Calcd for $C_{22}H_{21}NO_3S_2$: C, 64.21; H, 5.14; N, 3.40. Found: C, 63.94; H, 5.24; N, 3.26.

Minor isomer **23aB**: white powder (from EtOAc-hexane), mp 158–161 °C; 1H NMR ($CDCl_3$) δ : 1.48 (3 H, d, $J = 6.8$ Hz), 4.17 (1 H, d, $J = 4.9$ Hz), 4.31 (1 H, q, $J = 6.8$ Hz), 4.93 (1 H, d, $J = 4.9$ Hz), 7.09 (2 H, d, $J = 7$ Hz), 7.20–7.30 (8 H, m), 7.47–7.55 (3 H, m), 7.70 (2 H, d, $J = 8$ Hz); ^{13}C NMR ($CDCl_3$) δ : 21.2, 51.5, 57.7, 93.3, 125.6, 126.8, 127.1, 128.2, 128.7, 129.1, 129.3, 129.7, 132.7, 135.3, 139.0, 139.6; MS (FAB) m/z (rel int %): 412 (21, $M^+ + 1$), 154 (100); IR ν_{max} (KBr) cm^{-1} : 1335, 1170 (SO_2), 1050 (SO). Anal. Calcd for $C_{22}H_{21}NO_3S_2$: C, 64.21; H, 5.14; N, 3.40. Found: C, 64.08; H, 5.21; N, 3.22.

The Pummerer Reaction of 4-Sulfinyl- β -sultams 4, 19, and 23. General Procedure. To a solution of a 4-sulfinyl- β -sultam **4**, **19**, or **23** (0.2 mmol) in CH_2Cl_2 (1 cm^3) was added TFAA (113 μdm^3 , 0.8 mmol) at room temperature. After 20 h, CH_2Cl_2 (6 cm^3) and 1% aqueous NH_3 (6 cm^3) were added to the reaction mixture, and the organic layer was separated. The water layer was extracted once with CH_2Cl_2 (10 cm^3). The organic layer and the extracts were combined, washed with saturated aqueous NaCl (10 cm^3), dried ($MgSO_4$), and evaporated under reduced pressure. The residue was purified by preparative TLC with EtOAc-hexane (1:10–1:5 v/v).

S-Phenyl 2-(cyclohexylamino)-*p*-tolylthioacetate (7d): colorless oil; 1H NMR ($CDCl_3$) δ : 1.11–1.25 (5H, m), 1.62–1.70 (4 H, m), 1.96 (2 H, m), 2.34 (3 H, s), 2.54 (1 H, m), 4.63 (1 H, s), 7.17 (2 H, d, $J = 8.3$ Hz), 7.32 (2 H, d, $J = 8.3$ Hz), 7.35 (5 H, s); ^{13}C NMR ($CDCl_3$) δ : 21.1, 24.9, 25.0, 26.0, 33.4, 33.6, 55.2, 69.7, 127.5, 128.7, 129.0, 129.5, 134.4, 135.6, 138.0, 201.4, an aromatic carbon is overlapped; MS (FAB) m/z (rel int %): 340 (26, $M^+ + 1$), 154 (100); IR ν_{max} (NaCl) cm^{-1} : 3340 (NH), 1690 (CO). Anal. Calcd for $C_{21}H_{25}NOS$: C, 74.30; H, 7.42; N, 4.13. Found: C, 74.27; H, 7.50; N, 4.09.

S-Phenyl 2-(*N*-cyclohexyltrifluoroacetamido)-*p*-tolylthioacetate (8d): colorless oil; 1H NMR ($CDCl_3$) δ : 1.06–1.09 (1 H, m), 1.24–1.43 (3 H, m), 1.56–1.75 (4 H, m), 1.89 (1 H, br d, $J = 13$ Hz), 2.10 (1 H, br d, $J = 12$ Hz), 2.39 (3 H, s), 3.92 (1 H, m), 4.94 (1 H, s), 7.21 (2 H, d, $J = 7.8$ Hz), 7.35–7.41 (7 H, m); MS (FAB) m/z (rel int %): 436 (19, $M^+ + 1$), 326 (100); IR ν_{max} (NaCl) cm^{-1} : 1685 (CO), 1205, 1140 (CF_3). Anal. Calcd for $C_{23}H_{24}F_3NO_2S$: C, 63.43; H, 5.55; N, 3.22. Found: C, 63.17; H, 5.72; N, 3.09.

S-Phenyl 2-(cyclohexylamino)-3,3-dimethylbutane-thioate (7f): white solid (from hexane), mp 40–42 °C; 1H NMR ($CDCl_3$) δ : 1.03 (9 H, s), 1.00–1.31 (5 H, m), 1.51 (1 H, br s), 1.59–1.62 (1 H, m), 1.72–1.75 (2 H, m), 1.85 (1 H, br d, $J = 12$ Hz), 1.93 (1 H, br d, $J = 12$ Hz), 2.46–2.53 (1 H, m), 3.19 (1 H, s), 7.36–7.42 (5 H, m); ^{13}C NMR ($CDCl_3$) δ : 24.6, 25.0, 26.1, 27.2, 32.8, 34.4, 34.4, 55.6, 73.9, 129.0, 129.0, 129.4, 134.3, 203.5; MS (FAB) m/z (rel int %): 306 (21, $M^+ + 1$), 154 (100); IR ν_{max} (KBr) cm^{-1} : 3360 (NH), 1680 (CO). Anal. Calcd for $C_{18}H_{27}NOS$: C, 70.77; H, 8.91; N, 4.59. Found: C, 70.60; H, 8.91; N, 4.47.

(2R,1'R)-S-Phenyl 2-(α -methylbenzylamino)phenylthioacetate (20a): light yellow oil; 1H NMR ($CDCl_3$) δ : 1.43 (3 H, d, $J = 7$ Hz), 2.22 (1 H, br s), 3.98 (1 H, q, $J = 7$ Hz), 4.35 (1 H, s), 7.27–7.35 (15 H, m); ^{13}C NMR ($CDCl_3$) δ : 24.4, 57.0, 70.1, 126.7, 127.3, 127.4, 128.2, 128.6, 128.8, 128.8, 129.0, 129.1, 134.4, 138.1, 144.4, 201.3; MS (FAB) m/z (rel int %): 348 (38, $M^+ + 1$), 105 (100); IR ν_{max} (NaCl) cm^{-1} : 3345 (NH), 1695 (CO). Anal. Calcd for $C_{22}H_{21}NOS$: C, 76.05; H, 6.09; N, 4.03. Found: C, 76.12; H, 6.19; N, 4.03.

(2R,1'S)-S-Phenyl 2-(α -methylbenzylamino)-3,3-dimethylbutanethioate (20b): light yellow oil; 1H NMR ($CDCl_3$) δ : 1.06 (9 H, s), 1.39 (3 H, d, $J = 6.4$ Hz), 1.98 (1 H, br s), 3.22 (1 H, s), 3.90 (1 H, q, $J = 6.4$ Hz), 7.23–7.40 (10 H, m); ^{13}C NMR ($CDCl_3$) δ : 22.5, 27.1, 34.9, 57.0, 74.3, 126.9, 127.1, 128.4, 128.6, 129.0, 129.1, 134.3, 145.8, 202.0; MS (FAB) m/z (rel int %): 328 (64, $M^+ + 1$), 190 (100); IR ν_{max} (KBr) cm^{-1} : 3340 (NH), 1695 (CO). Anal. Calcd for $C_{20}H_{25}NOS$: C, 73.35; H, 7.69; N, 4.28. Found: C, 73.45; H, 7.83; N, 4.25.

(2S,1'R)-S-Phenyl 2-(α -methylbenzylamino)phenylthioacetate (24a): light yellow oil; 1H NMR ($CDCl_3$) δ : 1.38 (3 H, d, $J = 7$ Hz), 2.42 (1 H, br s), 3.70 (1 H, q, $J = 7$ Hz), 4.36 (1 H, s), 7.27–7.41 (15 H, m); ^{13}C NMR ($CDCl_3$) δ : 24.2, 55.2, 70.1, 127.0, 127.3, 128.0, 128.2, 128.4, 128.6, 128.8, 129.0, 129.2, 134.5, 137.8, 144.2, 199.3; MS (FAB) m/z (rel int %): 348 (51, $M^+ + 1$), 210 (100); IR ν_{max} (NaCl) cm^{-1} : 3340 (NH), 1700 (CO). Anal. Calcd for $C_{22}H_{21}NOS$: C, 76.05; H, 6.09; N, 4.03. Found: C, 76.14; H, 6.16; N, 4.08.

(2S,1'S)-S-Phenyl 2-(α -methylbenzylamino)-3,3-dimethylbutanethioate (24b): light yellow oil; 1H NMR ($CDCl_3$) δ : 0.97 (9 H, s), 1.35 (3 H, d, $J = 6.4$ Hz), 2.01 (1 H, br s), 2.91 (1 H, s), 3.88 (1 H, q, $J = 6.4$ Hz), 7.28 (1 H, t, $J = 7$ Hz), 7.34 (2 H, t, $J = 7$ Hz), 7.39 (2 H, d, $J = 7$ Hz), 7.42 (5 H, s); ^{13}C NMR ($CDCl_3$) δ : 25.3, 27.1, 34.3, 56.7, 74.0, 127.1, 127.2, 128.3, 128.6, 129.1, 129.2, 134.2, 144.8, 203.0; MS (FAB) m/z (rel int %): 328 (31, $M^+ + 1$), 190 (100); IR ν_{max} (KBr) cm^{-1} : 3345 (NH), 1695 (CO). Anal. Calcd for $C_{20}H_{25}NOS$: C, 73.35; H, 7.69; N, 4.28. Found: C, 73.35; H, 7.85; N, 4.26.

Acknowledgment. This work was supported in part by Grants-in-Aid for Scientific Research No. 08304037 and 09771915 from the Ministry of Education, Science, Sports, and Culture.

Supporting Information Available: Spectra data (1H NMR, ^{13}C NMR, IR, and MS) and assignment of 1H NMR signals for **1b**, **1c**, **2**, **4**, **5**, **7**, **8**, **14**, **17–24** and X-ray crystallographic analysis data for **14h** (41 pages). This material is contained in libraries on microfiche, immediately follows this article in the microfilm version of the journal, and can be ordered from the ACS; see any current masthead page for ordering information.

JO981230N